

OzCrop

Chemwatch: 5340-03

Version No: 4.1 Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements Chemwatch Hazard Alert Code: 2 Issue Date: 03/09/2020

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SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier

Product name	OzCrop Propaz Herbicide
Chemical Name	Not Applicable
Synonyms	APVMA Code: 66883
Proper shipping name	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains propyzamide)
Chemical formula	Not Applicable
Other means of identification	Not Available

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Agricultural herbicide
Neievant lucittineu uses	Agricultural herbicide.

Details of the manufacturer or supplier of the safety data sheet

Registered company name	OzCrop
Address	G13/25 Solent Circuit Norwest NSW 2153 Australia
Telephone	+61 2 8123 0170
Fax	+61 2 8123 0171
Website	http://www.ozcrop.com.au
Email	orders@ozcrop.com.au

Emergency telephone number

Association / Organisation	In Transport Emergency DIAL 000
Emergency telephone numbers	1800 033 111 (24 hours - Australia wide)
Other emergency telephone numbers	Not Available

SECTION 2 Hazards identification

Classification of the substance or mixture

Poisons Schedule	S6
Classification ^[1]	Serious Eye Damage/Eye Irritation Category 2B, Carcinogenicity Category 2, Hazardous to the Aquatic Environment Long-Term Hazard Category 1, Skin Corrosion/Irritation Category 2
Legend:	1. Classified by Chernwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

Label elements

Signal word Warning

Hazard statement(s)

Hazard

H320	Causes eye irritation.
H351	Suspected of causing cancer.
H410	Very toxic to aquatic life with long lasting effects.
H315	Causes skin irritation.

P201	Obtain special instructions before use.
P280	Wear protective gloves and protective clothing.
P273	Avoid release to the environment.
P264	Wash all exposed external body areas thoroughly after handling.
P264	Wash all exposed external body areas thoroughly after handling.

Precautionary statement(s) Response

P308+P313	IF exposed or concerned: Get medical advice/ attention.
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P337+P313	If eye irritation persists: Get medical advice/attention.
P391	Collect spillage.
P302+P352	IF ON SKIN: Wash with plenty of water.
P332+P313	If skin irritation occurs: Get medical advice/attention.
P362+P364	Take off contaminated clothing and wash it before reuse.

Precautionary statement(s) Storage

P405 Store locked up.

Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
23950-58-5	30-60	propyzamide
Not Available		(500g/L)
107-21-1	1-10	ethylene glycol
Not Available		(42g/L)
Not Available	<10	Ingredients determined not to be hazardous
7732-18-5	balance	water
Legend:	1. Classified by Chemwatch; 2. C Classification drawn from C&L * I	assification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. EU IOELVs available

SECTION 4 First aid measures

Description of first aid measures		
Eye Contact	 If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel. 	
Skin Contact	 If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation. 	
Inhalation	 If fumes, aerosols or combustion products are inhaled remove from contaminated area. Other measures are usually unnecessary. 	
Ingestion	 For advice, contact a Poisons Information Centre or a doctor at once. Urgent hospital treatment is likely to be needed. If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Transport to hospital or doctor without delay. 	

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

- For acute or short term repeated exposures to ethylene glycol:
- Early treatment of ingestion is important. Ensure emesis is satisfactory.
- Test and correct for metabolic acidosis and hypocalcaemia.
- Apply sustained diuresis when possible with hypertonic mannitol.
- Evaluate renal status and begin haemodialysis if indicated. [I.L.O]
- Rapid absorption is an indication that emesis or lavage is effective only in the first few hours. Cathartics and charcoal are generally not effective.
- Correct acidosis, fluid/electrolyte balance and respiratory depression in the usual manner. Systemic acidosis (below 7.2) can be treated with intravenous sodium bicarbonate solution.
- Ethanol therapy prolongs the half-life of ethylene glycol and reduces the formation of toxic metabolites.

- Pyridoxine and thiamine are cofactors for ethylene glycol metabolism and should be given (50 to 100 mg respectively) intramuscularly, four times per day for 2 days.
 Magnesium is also a cofactor and should be replenished. The status of 4-methylpyrazole, in the treatment regime, is still uncertain. For clearance of the material and its
- metabolites, haemodialysis is much superior to peritoneal dialysis.
- [Ellenhorn and Barceloux: Medical Toxicology]

It has been suggested that there is a need for establishing a new biological exposure limit before a workshift that is clearly below 100 mmol ethoxy-acetic acids per mole creatinine in morning urine of people occupationally exposed to ethylene glycol ethers. This arises from the finding that an increase in urinary stones may be associated with such exposures. *Laitinen J., et al: Occupational & Environmental Medicine* 1996; 53, 595-600

SECTION 5 Firefighting measures

Extinguishing media

The product contains a substantial proportion of water, therefore there are no restrictions on the type of extinguishing media which may be used. Choice of extinguishing media should take into account surrounding areas.

Though the material is non-combustible, evaporation of water from the mixture, caused by the heat of nearby fire, may produce floating layers of combustible substances. In such an event consider:

- foam.
- dry chemical powder.
- carbon dioxide.

Special hazards arising from the substrate or mixture

Fire Incompatibility	None known.

Advice for firefighters

Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves in the event of a fire. Prevent, by any means available, spillage from entering drains or water courses. Use fire fighting procedures suitable for surrounding area. DO NOT approach containers suspected to be hot. Cool fire exposed containers with water spray from a protected location. If safe to do so, remove containers from path of fire. Equipment should be thoroughly decontaminated after use.
Fire/Explosion Hazard	 The material is not readily combustible under normal conditions. However, it will break down under fire conditions and the organic component may burn. Not considered to be a significant fire risk. Heat may cause expansion or decomposition with violent rupture of containers. Decomposes on heating and may produce toxic fumes of carbon monoxide (CO). May emit acrid smoke. Decomposes on heating and produces toxic fumes of: carbon dioxide (CO2) hydrogen chloride phosgene nitrogen oxides (NOX)
HAZCHEM	other pyrolysis products typical of burning organic material. •3Z

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures See section 8

Environmental precautions See section 12

Methods and material for containment and cleaning up

Minor Spills	 Environmental hazard - contain spillage. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite. Wipe up. Place in a suitable, labelled container for waste disposal.
Major Spills	 Environmental hazard - contain spillage. Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves. Prevent, by any means available, spillage from entering drains or water course. Stop leak if safe to do so. Contain spill with sand, earth or vermiculite. Collect recoverable product into labelled containers for recycling. Neutralise/decontaminate residue (see Section 13 for specific agent). Collect solid residues and seal in labelled drums for disposal. Wash area and prevent runoff into drains. After clean up operations, decontaminate and launder all protective clothing and equipment before storing and re-using. If contamination of drains or waterways occurs, advise emergency services.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

Safe handling	 DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Avoid contact with moisture. Avoid contact with moisture. Avoid contact with incompatible materials. When handling, DO NOT eat, drink or smoke. Keep containers securely sealed when not in use. Avoid physical damage to containers. Always wash hands with soap and water after handling. Work clothes should be laundered separately. Launder contaminated clothing before re-use. Use good occupational work practice. Observe manufacturer's storage and handling recommendations contained within this SDS. Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.
Other information	 Store in original containers. Keep containers securely sealed. Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers. Protect containers against physical damage and check regularly for leaks. Observe manufacturer's storage and handling recommendations contained within this SDS.

Conditions for safe storage, including any incompatibilities

Suitable container	 Polyethylene or polypropylene container. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks.
Storage incompatibility	None known

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	ethylene glycol	Ethylene glycol (vapour)	20 ppm / 52 mg/m3	104 mg/m3 / 40 ppm	Not Available	Not Available
Australia Exposure Standards	ethylene glycol	Ethylene glycol (particulate)	10 mg/m3	Not Available	Not Available	Not Available

Emergency Limits

Ingredient	TEEL-1	TEEL-2		TEEL-3
ethylene glycol	30 ppm	150 ppm		900 ppm
Ingredient	Original IDLH		Revised IDLH	
propyzamide	Not Available		Not Available	
ethylene glycol	Not Available		Not Available	
water	Not Available		Not Available	

Occupational Exposure Banding			
Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit	
propyzamide	E	≤ 0.01 mg/m³	
Notes:	Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a range of exposure concentrations that are expected to protect worker health.		

Exposure controls

	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and vern "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed proper ventilation system must match the particular process and chemical or contaminant in use. Employers may need to use multiple types of controls to prevent employee overexposure.	engineering controls can of protection. ntilation that strategically rly. The design of a
Appropriate engineering controls	General exhaust is adequate under normal operating conditions. Local exhaust ventilation may be required in specific ci overexposure exists, wear approved respirator. Correct fit is essential to obtain adequate protection. Provide adequate v or closed storage areas. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn velocities" of fresh circulating air required to effectively remove the contaminant.	rcumstances. If risk of rentilation in warehouse n, determine the "capture
	Type of Contaminant:	Air Speed:
	solvent, vapours, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50-100 f/min)
	aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)	0.5-1 m/s (100-200 f/min.)

	direct spray, spray painting in shallow booths, drum filling, generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)			
	grinding, abrasive blasting, tumbling, high speed wheel gen very high rapid air motion).	2.5-10 m/s (500-2000 f/min.)			
	Within each range the appropriate value depends on:				
	Lower end of the range	Upper end of the range			
	1: Room air currents minimal or favourable to capture	1: Disturbing room air currents			
	2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity			
	3: Intermittent, low production.	3: High production, heavy use			
	4: Large hood or large air mass in motion	4: Small hood-local control only			
	Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.				
Personal protection					
Eye and face protection	 Safety glasses with side shields. Chemical goggles. Contact lenses may pose a special hazard; soft contact lenses may pose a special hazard; soft contact lenses may pose or restrictions on use, should be craand adsorption for the class of chemicals in use and an a their removal and suitable equipment should be readily a remove contact lens as soon as practicable. Lens should a clean environment only after workers have washed har national equivalent] 	lenses may absorb and concentrate irritants. A written policy reated for each workplace or task. This should include a revi account of injury experience. Medical and first-aid personnel available. In the event of chemical exposure, begin eye irriga d be removed at the first signs of eye redness or irritation - le nds thoroughly. [CDC NIOSH Current Intelligence Bulletin 59	v document, describing lew of lens absorption I should be trained in tition immediately and ens should be removed in 9], [AS/NZS 1336 or		
Skin protection	See Hand protection below				
Hands/feet protection	 Wear safety footwear or safety gumboots, e.g. Rubber The selection of suitable gloves does not only depend on the manufacturer. Where the chemical is a preparation of severa and has therefore to be checked prior to the application. The exact break through time for substances has to be obtain making a final choice. Personal hygiene is a key element of effective hand care. Glo washed and dried thoroughly. Application of a non-perfumed Suitability and durability of glove type is dependent on usage i frequency and duration of contact, chemical resistance of glove material, glove thickness and dexterity Select gloves tested to a relevant standard (e.g. Europe EN 3 When prolonged or frequently repeated contact may occur, minutes according to EN 374, AS/NZS 2161.10.1 or national When only brief contact is expected, a glove with a protection 374, AS/NZS 2161.10.1 or national equivalent) is recomment contaminated gloves should be replaced. As defined in ASTM F-739-96 in any application, gloves are if Excellent when breakthrough time > 20 min Fair when breakthrough time > 20 min Foor when glove material degrades For general applications, gloves with a thickness typically great it should be emphasised that glove thickness is not necessar efficiency of the glove will be dependent on the exact compoor consideration of the task requirements and knowledge of break Glove thickness may also vary depending on the glove manu data should always be taken into account to ensure selection Note: Depending on the activity being conducted, gloves of v . Thinner gloves (down to 0.1 mm or less) may be required when puncture potential Gloves must only be worn on clean hands. After using gloves 	e material, but also on further marks of quality which vary fro I substances, the resistance of the glove material can not be ned from the manufacturer of the protective gloves and has oves must only be worn on clean hands. After using gloves, moisturiser is recommended. a Important factors in the selection of gloves include: 374, US F739, AS/NZS 2161.1 or national equivalent). a glove with a protection class of 5 or higher (breakthrough equivalent) is recommended. on class of 3 or higher (breakthrough time greater than 60 m ded. and this should be taken into account when considering glover rated as: eater than 0.35 mm, are recommended. rily a good predictor of glove resistance to a specific chemic sition of the glove material. Therefore, glove selection should akthrough times. fracturer, the glove type and the glove model. Therefore, the of of the most appropriate glove for the task. rarying thickness may be required for specific tasks. For exa where a high degree of manual dexterity is needed. However just for single use applications, then disposed of. e there is a mechanical (as well as a chemical) risk i.e. where s, hands should be washed and dried thoroughly. Application	m manufacturer to e calculated in advance to be observed when hands should be time greater than 240 ninutes according to EN res for long-term use. al, as the permeation d also be based on e manufacturers technical mple: , these gloves are only re there is abrasion or n of a non-perfumed		
	Groves must only be worn on clean nanos. After using gloves, nanos should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.				
Body protection	See Other protection below				
Other protection	 Overalls. P.V.C apron. Barrier cream. Skin cleansing cream. Five weak with 				

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001,

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

OzCrop Propaz Herbicide

Material	СРІ
BUTYL	C
NATURAL RUBBER	С
NATURAL+NEOPRENE	С
NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
NITRILE+PVC	С
PE/EVAL/PE	С
PVA	С
PVC	С
TEFLON	С
VITON	С

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	A-AUS P2	-	A-PAPR-AUS / Class 1 P2
up to 50 x ES	-	A-AUS / Class 1 P2	-
up to 100 x ES	-	A-2 P2	A-PAPR-2 P2 ^

^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

Appearance	Light brown coloured liquid with mild odour; mixes with water.		
Physical state	Liquid	Relative density (Water = 1)	1.1-1.2 @20C
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable
pH (as supplied)	Neutral	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	<0	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	~100	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	*2.37 @20C (water)	Gas group	Not Available
Solubility in water	Miscible	pH as a solution (Not Available%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Information on toxicological ef	ffects			
Inhaled	The material is not thought to produce either adverse health effects or irr Directives using animal models). Nevertheless, adverse systemic effects route and good hygiene practice requires that exposure be kept to a min setting.	ritation of the respiratory tract following inhalation (as classified by EC s have been produced following exposure of animals by at least one other imum and that suitable control measures be used in an occupational		
Ingestion	Accidental ingestion of the material may be damaging to the health of the	e individual.		
Skin Contact	This material can cause inflammation of the skin on contact in some persons. The material may accentuate any pre-existing dermatitis condition Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.			
Eye	There is some evidence to suggest that this material can cause eye irrita	ation and damage in some persons.		
Chronic	There has been concern that this material can cause cancer or mutations, but there is not enough data to make an assessment. There is some evidence from animal testing that exposure to this material may result in reduced fertility. There is some evidence from animal testing that exposure to this material may result in toxic effects to the unborn baby. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. Exposure to the material for prolonged periods may cause physical defects in the developing embryo (teratogenesis).			
	τοχιείτη	IRRITATION		
OzCrop Propaz Herbicide	Not Available	Not Available		
	ΤΟΧΙΟΙΤΥ	IRRITATION		
propyzamide	Dermal (rabbit) LD50: >3160 mg/kg ^[2]	Eye: slight		
	Oral (Rat) LD50; 5620 mg/kg ^[2]	Skin: slight		
	τοχιζιτγ	IRRITATION		
	dermal (mouse) LD50: >3500 mg/kg ^[1]	Eye (rabbit): 100 mg/1h - mild		
	Oral (Bat) D50: >2000 mo/ko ^[2]	Eve (rabbit): 12 mg/m3/3D		
		Eve (rabbit): 1440mo/6h-moderate		
ethylene glycol		Eve (rabbit): 500 mo/24h - mild		
		Eve: no adverse effect observed (not irritatino) ^[1]		
		Skin (rabbit): 555 mg(open)-mild		
		Skin: no adverse effect observed (not irritating) ^[1]		
	ΤΟΧΙΟΙΤΥ	IRRITATION		
water	Oral (Rat) LD50; >90000 mg/kg ^[2]	Not Available		
Legend:	 Value obtained from Europe ECHA Registered Substances - Acute to specified data extracted from RTECS - Register of Toxic Effect of chemin 	xicity 2. Value obtained from manufacturer's SDS. Unless otherwise cal Substances		
	·			
PROPYZAMIDE	NOEL: (chronic studies) dogs 300, rats 200, mice 13 mg/kg diet * ADI 0. For propyzamide (pronamide): Pronamide is practically non-toxic via swallowing and is slightly toxic thro cause changes to thyroid, adrenal gland, and pituitary function, with deci- testing at sufficient doses. It is unlikely to cause reproductive toxicity, birl at sufficient doses, notably of the liver. Microtubules (MTs) are hollow cylindrical polymers composed of alpha-b cytoplasm during interphase and form the mitotic spindle to segregate cc a remarkable diversity in eukaryotes (organisms with a central cell nucle ancestors. While alpha and beta-tubulin are highly conserved proteins, the effects of evolutionary groups. For example, plant tubulin and Apicomplexan tubuli contrast, small synthetic molecules such as dinitroanilines (orzalin, etha but not vertebrate or fungi ones. Due to their selectivity towards plant tut and represent promising leads for the design of antiparasite drug candid Besides dinitroanilines and their derivatives, no chemical entities that se This is not the case for mammalian tubulin, which is the target of numero Whilst mammalian microtubules are resistant to dinitroaniline herbicides, homology. Molecular modeling of plant alpha/beta tubulin dimer indicates The resistance of plants to dinitroaniline herbicides (such as trifluralin an tubulin. The Thr239 residue is located near the end of the long central he beta-tubulin of the next dimer in the microtubule protofilament. Therefore provide a binding site for structurally unrelated alpha-tubulin inhibitors, a inhibitors. The microtubules of some protozoan species are also suscep. The tubulin from the protozoan Plasmdium falciparum contains a dinitroa- humans and might be a target for new antimalarial drugs.	08 mg/kg * Toxicity Class WHO Table 5; EPA IV * ough skin contact, causing slight local irritation. Chronic exposure can reases in food consumption and changes in blood chemistry in animal th defects or mutations. Data suggests that pronamide may cause cancer reta tubulin heterodimers. These highly dynamic assemblies organize the ondensed chromosomes during mitosis. Microtubule organization shows us), with striking differences in clades deriving from photosynthetic of microtubule-binding drugs vary in organisms belonging to distinct ins have a much lower affinity for colchicine than animal tubulin In afluralin or trifluralin) bind specifically to plant and Apicomplexa tubulins buin, dinitroanilines have been used as herbicides for more than 40 years lates in particular in the case of P. falciparum and T. gondii lectively target tubulin of plants and parasites have yet been described. bus diverse chemical compounds. , the sequences of alpha-tubulin from plants and animals show significant s a likely dinitroaniline binding site in the area of dimer-to-dimer contact. di oryzalin) has been found to be caused by a Thr239 mutation of alpha- elix H7, and is positioned close to the site that interacts with the e, alpha-tubulin residues near the interface of beta-tubulin appear to and some mutations in this region appear to lead to resistance to the bible to dirsuption by dinitroaniline herbicides aniline/phosphorothicamidate-binding site that is not conserved in ang center such as the centrosome, or in many differentiated animal cells		

	longitudinally oriented array of two-three sub-pellicula [* The Pesticides Manual, Incorporating The Agro Council]	ar microtubules contributes to the shap chemicals Handbook, 10th Edition,	be and integrity of the parasite Editor Clive Tomlin, 1994, British Crop Protection
ETHYLENE GLYCOL	[Estimated Lethal Dose (human) 100 ml; RTECS quot For ethylene glycol: Ethylene glycol is quickly and extensively absorbed th through the airways; absorption through skin is appare metabolized by alcohol dehydrogenase to form glycoa are oxidized to glyoxylate, which is one of the majo glycol is eliminated in the urine as both the parent con Respiratory effects: Respiratory system involvement of include hyperventilation, shallow rapid breathing, and the lungs. Respiratory system involvement appears to may be other changes compatible with adult respirato aspiration of stomach contents. Symptoms related to a symptoms such as swelling of the lung and inflammati poisoning. Cardiovascular effects: Cardiovascular system involves second phase of ethylene glycol poisoning by swallow heart include increased heart rate, heart enlargement to cardiogenic shock. In lethal cases, inflammation of rare and usually seen after swallowing higher doses oo serious cardiovascular effects: Reported musculoskeletal eff pain, associated with high levels of creatinine in the bl Liver effects: Adverse kidney effects are seen during oxalate crystals are deposited in the tubules and are are inflammation of the tubule interstitium. If untreated, the decreased kidney function, reduction in urine output a to normal or near normal. Metabolic effects: Metabolic changes can occur within accumulation of glycolic acid in the blood and therefor anions (mainly glycolate). Effects on the nervous system: Adverse reactions invo glycol is swallowed. These early effects are also the oo (see above), they occur from 0.5-12 hours after expos Inco-ordination, slurred speech, confusion and sleepin there may be effects: Animal testing showed that ethy Effects on development: Animal studies indicate that the weight. Cancer: No studies are known regarding cancer effect Genetic toxicity: No human studies available, but anim	ted by Orica] Substance is reproductiv roughout the gastrointestinal tract. Lir ently slow. Following absorption, it is of aldehyde, which is rapidly converted to bolized to formic acid, oxalic acid, and or elimination products of ethylene glyun npound and glycolic acid. Elimination occurs 12-24 hours after swallowing su generalized swelling of the lungs with be dose-dependent and occurs at the ry distress syndrome (ARDS). Swellin acidosis such as fast or excessive bre- ion of the bronchi and lungs are relative ement in humans occurs at the same the ving, which is 12-24 hours after acute and ventricular gallop. There may als the heart muscle has been observed i of ethylene glycol. In summary, acute ef- of a long-term, low-dose exposure are to favellowing ethylene glycol includer termittent diarrhea and pain, and after fects in cases of acute ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ell death (necrosis) of the liver. g the third stage of ethylene glycol pois- ser on the urine. There may also be co- the are reduction in blood pH. The anion the same and are considered to be part of thes are common in the early stages, reversible over many mon	re effector in rats (birth defects). Mutagenic to rat cells. mited information suggests that it is also absorbed distributed throughout the body. In humans, it is initially o glycolic acid and glyoxal. These breakdown products I glycine. Breakdown of both glycine and formic acid col. In addition to exhaled carbon dioxide, ethylene is rapid and occurs within a few hours. ufficient amounts of ethylene glycol. Symptoms realcium oxalate deposits occasionally appearing in e same time as cardiovascular changes. Later, there ag of the lung can be a result of heart failure, ARDS, or rathing are frequently observed; however, major vely rare, and are usually seen only in extreme time as respiratory system involvement, during the exposure. The symptoms of poisoning involving the o be high or low blood pressure, which may progress at autopsy. Cardiovascular involvement appears to be exposure to high levels of ethylene glycol can cause e unknown. ausea, vomiting with or without blood, heartburn and surgery, deposition of oxalate crystals was shown to poisoning include diffuse muscle tenderness and ciated with low calcium. oning showed deposition of calcium oxalate in the degeneration and death of tubule cells, and us and leads to blood and protein in the urine, quate supportive therapy, kidney function can return col. There may be metabolic acidosis, caused by gap is increased, due to increased unmeasured the first symptoms to appear in humans after ethylene sed ethylene glycol. Together with metabolic effects the first stage in ethylene glycol poisoning. as are irritation, restlessness and disorientation. Later, ng of the brain (cerebrum) and crystal deposits of people who died after acute ethylene glycol I of fetuses and the male reproductive organs. e in pregnancy; there may also be reduction in foetal osure to ethylene glycol. pative.
WATER	INO SIGNITICANT ACUTE TOXICOlOGICAL DATA IDENTIFIED IN LITER	rature search.	•
Acute Toxicity	×	Carcinogenicity	✓ Ŭ
Skin irritation/Corrosion	· · · · · · · · · · · · · · · · · · ·	Reproductivity	\$
Respiratory or Skin sensitisation	×	STOT - Repeated Exposure	x
Mutagenicity	×	Aspiration Hazard	×
		Legend: X – Data either r ✓ – Data availab	not available or does not fill the criteria for classification le to make classification

SECTION 12 Ecological information

	Endpoint	Test Duration (hr)	Species	Value	Source
OzCrop Propaz Herbicide	Not Available	Not Available	Not Available	Not Available	Not Availabl
	Endpoint	Test Duration (hr)	Species	Value	Source
propyzamide	EC50	48h	Crustacea	>5.6mg/L	4
	NOEC(ECx)	1008h	Algae or other aquatic plants	0.1mg/L	4
	LC50	96h	Fish	>7.69mg/L	4
	Endpoint	Test Duration (hr)	Species	Value	Sourc
ethylene glycol	EC50(ECx)	Not Available	Algae or other aquatic plants	6500-7500mg/l	1
	EC50	48h	Crustacea	>100mg/l	2

	LC50	96h	Fish	>100	000mg/l	1
	EC50	96h	Algae or other aquatic plants	6500	0-13000mg/l	1
	Endpoint	Test Duration (hr)	Species		Value	Source
water	Not Available	Not Available	Not Available		Not Available	Not Available
Legend:	Extracted from Ecotox databas - Bioconcentra	1. IUCLID Toxicity Data 2. Europe ECHA Registe se - Aquatic Toxicity Data 5. ECETOC Aquatic Ha tion Data 8. Vendor Data	red Substances - Ecotoxicological Information - zard Assessment Data 6. NITE (Japan) - Biocon	Aquati icentra	c Toxicity 4. L tion Data 7. M	JS EPA, IETI (Japan)

Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment. **DO NOT** discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
propyzamide	HIGH	HIGH
ethylene glycol	LOW (Half-life = 24 days)	LOW (Half-life = 3.46 days)
water	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation
propyzamide	LOW (LogKOW = 3.5747)
ethylene glycol	LOW (BCF = 200)

Mobility in soil

Ingredient	Mobility
propyzamide	LOW (KOC = 1587)
ethylene glycol	HIGH (KOC = 1)

SECTION 13 Disposal considerations

Waste treatment methods	
Product / Packaging disposal	 Containers may still present a chemical hazard/ danger when empty. Return to supplier for reuse/ recycling if possible. Otherwise: If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill. Where possible retain label warnings and SDS and observe all notices pertaining to the product. DO NOT allow wash water from cleaning or process equipment to enter drains. It may be necessary to collect all wash water for treatment before disposal. In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first. Where in doubt contact the responsible authority. Recycle wherever possible or consult manufacturer for recycling options. Consult State Land Waste Authority for disposal. Bury or incinerate residue at an approved site. Recycle containers if possible, or dispose of in an authorised landfill.

SECTION 14 Transport information

Labels Required

Marine Pollutant	
HAZCHEM	•3Z

Land transport (ADG)

UN number	3082		
UN proper shipping name	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains propyzamide)		
Transport hazard class(es)	Class Subrisk	9 Not Applicable	

Packing group	Ш		
Environmental hazard	Environmentally hazardous		
Special precautions for user	Special provisions	274 331 335 375 AU01	
	Limited quantity	5L	

Environmentally Hazardous Substances meeting the descriptions of UN 3077 or UN 3082 are not subject to this Code when transported by road or rail in; (a) packagings;

(b) IBCs; or

(c) any other receptacle not exceeding 500 kg(L). - Australian Special Provisions (SP AU01) - ADG Code 7th Ed.

Air transport (ICAO-IATA / DGR)

UN number	3082			
UN proper shipping name	Environmentally hazardous substance, liquid, n.o.s. * (contains propyzamide)			
Transport hazard class(es)	ICAO/IATA Class ICAO / IATA Subrisk ERG Code	9 Not Applicable 9L		
Packing group	II			
Environmental hazard	Environmentally hazardo	bus		
Special precautions for user	Special provisions Cargo Only Packing Instructions Cargo Only Maximum Qty / Pack Passenger and Cargo Packing Instructions Passenger and Cargo Maximum Qty / Pack Passenger and Cargo Limited Quantity Packing Instructions Passenger and Cargo Limited Maximum Qty / Pack		A97 A158 A197 A215 964 450 L 964 450 L 450 L Y964 30 kg G	· · ·

Sea transport (IMDG-Code / GGVSee)

UN number	3082		
UN proper shipping name	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains propyzamide)		
Transport hazard class(es)	IMDG Class IMDG Subrisk	9 Not Applicable	
Packing group	III		
Environmental hazard	Marine Pollutant		
Special precautions for user	EMS Number Special provisions Limited Quantities	F-A, S-F 274 335 969 5 L	

Transport in bulk according to Annex II of MARPOL and the IBC code Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
propyzamide	Not Available
ethylene glycol	Not Available
water	Not Available

Transport in bulk in accordance with the ICG Code

Product name	Ship Type
propyzamide	Not Available
ethylene glycol	Not Available
water	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

propyzamide is found on the following regulatory lists

Australia Chemicals with non-industrial uses removed from the Australian Inventory of Chemical Substances (old Inventory)

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -Schedule 5

Chemical Footprint Project - Chemicals of High Concern List

Continued...

ethylene glycol is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 10 / Appendix C

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 5

water is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

National Inventory Status

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -Schedule 6 Australian Inventory of Industrial Chemicals (AIIC)

Chemical Footprint Project - Chemicals of High Concern List

National Inventory	Status
Australia - AIIC / Australia Non-Industrial Use	Yes
Canada - DSL	No (propyzamide)
Canada - NDSL	No (propyzamide; ethylene glycol; water)
China - IECSC	No (propyzamide)
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	No (propyzamide)
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	No (propyzamide)
USA - TSCA	No (propyzamide)
Taiwan - TCSI	Yes
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - FBEPH	No (propyzamide)
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

SECTION 16 Other information

Revision Date	03/09/2020
Initial Date	27/12/2018

SDS Version Summary

Version	Date of Update	Sections Updated
3.1	01/11/2019	One-off system update. NOTE: This may or may not change the GHS classification
4.1	03/09/2020	Classification change due to full database hazard calculation/update.

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit。 IDLH: Immediately Dangerous to Life or Health Concentrations ES: Exposure Standard OSF: Odour Safety Factor NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors **BEI: Biological Exposure Index** AIIC: Australian Inventory of Industrial Chemicals DSL: Domestic Substances List NDSL: Non-Domestic Substances List IECSC: Inventory of Existing Chemical Substance in China EINECS: European INventory of Existing Commercial chemical Substances ELINCS: European List of Notified Chemical Substances NLP: No-Longer Polymers ENCS: Existing and New Chemical Substances Inventory KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals

PICCS: Philippine Inventory of Chemicals and Chemical Substances TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas NCI: National Chemical Inventory FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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